

What's the Functional Solubility?

The landscape, challenges and current limitations faced when working with 'difficult-to-test' substances in aquatic matrices.

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Regulatory Requirements

Aquatic Ecotoxicology Testing

- Multiple treatment levels required by OECD/OCSP guidelines
- Analytical recovery requirements 80-120% of nominal or $\pm 20\%$ of mean measured concentrations.
- OECD, 2000 and USEPA, 1996 define “poorly soluble” as solubility limit < 100 mg/L



Physico-Chemical Properties

Impacting Test Substance Behavior In Water:

- Solubility
- Kow
- Kd
- Hydrolysis
- Photolysis
- Oxidation
- Biodegradation
- Volatility



Additional Contributors

- Matrix-Related Factors
 - pH
 - Total Organic Carbon
 - Water Hardness
 - Salinity
- Water can vary between labs – all freshwater is not the same
- It's important to note that **functional solubility within an exposure system** is often quite different than reported solubility (e.g., shake-flask)



Examples of 'Difficult-to-Test' Substances



- Pesticide AI's and formulations



- Pharmaceuticals

- Personal Care Products



- Petroleum distillates

- Complex mixtures



- UVCBs



Delivery Methodologies

- The use of carrier solvents, up to 100 $\mu\text{L}/\text{L}$, is acceptable under the guidelines, but there are potential confounding issues resulting from their use, especially during chronic exposures.
 - Biofilms
 - Statistical comparisons

Non-solvent options for delivery:

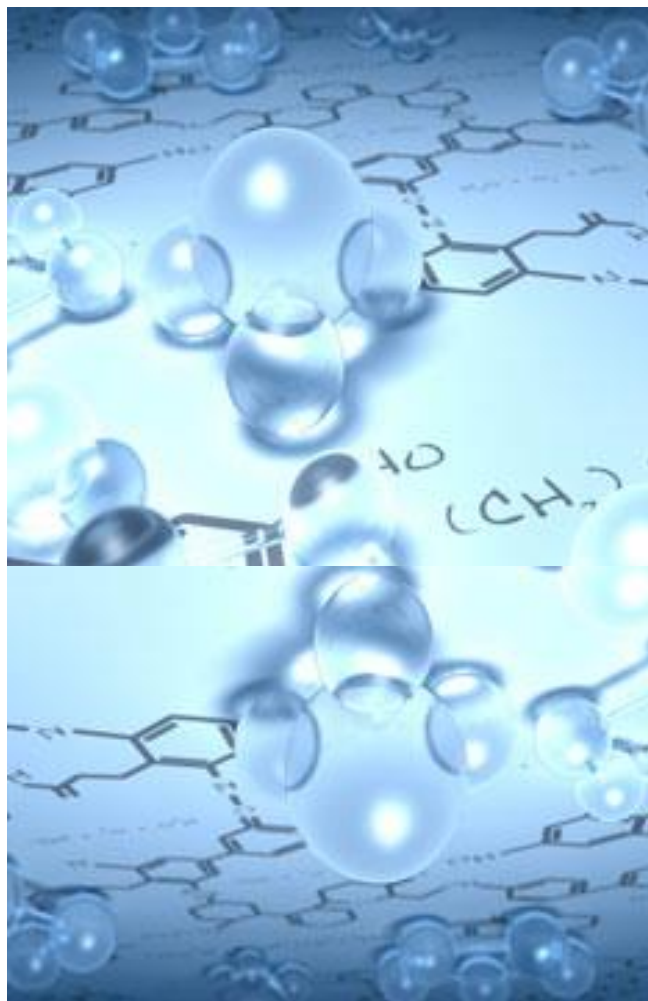
- Direct addition
- Water Accommodated Fraction (WAF)
- Saturator systems:
 - Glass wool saturator columns
 - Solid-liquid saturator (slow-stir)
 - Liquid-liquid saturator (slow-stir)
 - Passive dosing methods
- Each of these methods has limitations and can require resources to explore options



The image shows two identical 100 ml volumetric flasks filled with an orange liquid. The flasks are positioned side-by-side against a blurred background of light green and yellow. A semi-transparent blue horizontal bar is centered across the middle of the image, containing the text 'CASE STUDY' in white, bold, sans-serif capital letters. The flasks have black stoppers and are marked with '100 ml' and '20°C' on their bodies.

CASE STUDY

Model Compound – broad-spectrum fungicide



Test Substance Characteristics

Physical State: Solid Powder

Water Solubility: 1.8 mg/L (low solubility)

Solvent Solubility: 190 g/L in acetone

Log Pow: 4.14

pKa:
pKa1 = < 0 (acidic)
pKa2 = 14 (basic)

Aqueous Hydrolysis: Stable at pH 5, 7 and 9

Aqueous Photolysis: DT₅₀ = 10 days at pH 7

Vapor Pressure: 3.9 x10⁻⁷ Pa at 25°C (non-volatile)



Existing Data (1991 – 2011)

Study Type	Year	Solvent Type	Solvent Load ($\mu\text{L/L}$)	High Dose ($\mu\text{g/L}$)	Percent Recovery	Perceived Solvent Effects?	Repeat Required?
Daphnia Acute	1993	Acetone	1000	5000	38	No	No
Daphnia Life Cycle	1991	Acetone	17	320	59	Yes	Yes
FHM Early Life-Stage	1992	Acetone	18	300	90	Yes	Yes
FHM Early Life-Stage	1994	Acetone	280	150	96	Yes	Yes
SHM Early Life-Stage	2011	DMF	100	625	92	Yes	Yes



Evaluating Test Design & Generating New Data

Daphnia Magna Life-Cycle

- Data considered:
- Hydrolytically stable = semi-static design
- Toxicity and solubility data derived from previous studies
- Benchtop solubility in DI water
 - 1000 $\mu\text{g/L}$ nominal yielded 600 $\mu\text{g/L}$
- Highest recovery in test water = 35 $\mu\text{g/L}$
 - Test water = hard water
- Concentrations considerably less than water solubility (1800 $\mu\text{g/L}$)



Mysid & Fish Studies (2014)

Updated Delivery Method:

- Data considered:
 - Saturator column suitable for $\log K_{ow} > 2.75$
 - Column trial:
 - Measured **1800 $\mu\text{g/L}$** (freshwater)
1300 $\mu\text{g/L}$ (saltwater)
 - Column life: ≥ 30 days

Study Type	High Dose ($\mu\text{g/L}$)	Mean Measured Dose ($\mu\text{g/L}$)	Percent Recovery
Mysid Life Cycle	80	96	120
SHM Early Life Stage	300	330	110
FHM Life Cycle	90	86	96

- *Avoided solvent-related effects*



Lessons Learned

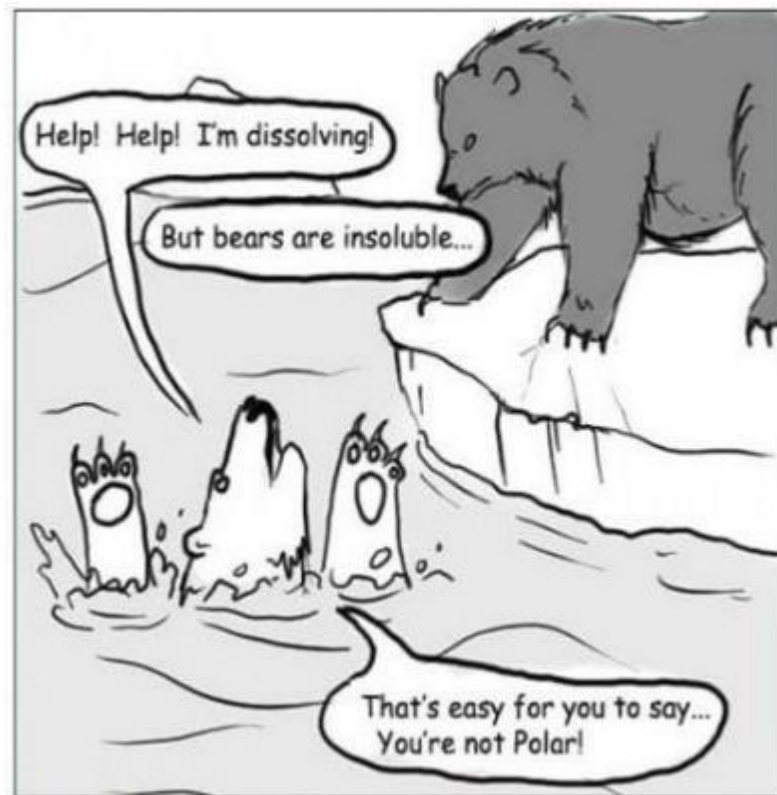
Using solvent may save time and money upfront, however...

- Solvent effects (*or the perception of effects*) can be difficult to avoid
- Costs are increased with repeating studies
 - Cost for repeating *Daphnia* and chronic fish studies for this DCI: **~\$120,000**
 - Cost for developing saturator column data: **~\$7,000**



Additional Considerations

- Limitations of non-solvent delivery
 - Some methods not practical for every compound
- Importance of benchtop lab trials
 - Can result in additional costs, but the trade-off is worthwhile
- When would a saturator column not be appropriate?
 - Some compounds can be challenging due to highly variable column output
 - Liquids, especially hydrocarbons are not good candidates, but other methodologies are available
- Sometimes solvent use is unavoidable, but use must be considered carefully



Thank You...
Questions?